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Introduction

Introduction The unique adaptability of the green alga Dunaliella to grow in a wide range of salt concentrations, has made it a favorite subject for detailed physiological and biochemical investigations, aimed at its utility as a source of some fine chemicals such as β -carotene and glycerol or a source of single cell protein for rearing rotifers and brine shrimp, Artemia. It is well known that the concentrations and the ratios of the essential nutritive elements, strongly affect algal growth and photosynthetic rates (FISHER et al., 1981). The main target of this work is to maximize Dunaliella salina growth with the most economical culture commencent supple component supply.

Aaterial and methods

Material and methods The methodology of experimental design, namely, the random balance, the fractional factorial and the steepest ascent designs (SATTERTHWAITE, 1959; COCHRAN and COX, 1957) were used to develop a new synthetic optimized medium for the growth of *Dunaliella salina*. All major cations (Na, K, Mg and Ca) were added as chloride /salts and anions (HCO3, SO4,NO3 and PO4) as sodium salts. The trace metals (Zn, Mn, Mo, Co, Cu and Fe) were added in chelated form with EDTA. Vitamin B₁₂ was added to all media at a level of L ug/l. Culture media were inoculated under sterile conditions with actively growing *Dunaliella salina*, adjusting its initial concentration to 104 cell/l. Experiments were performed in triplicates. Cultures were grown in incubator at light intensity 4 k Lux and temperature of 25 \pm 1°C. Experiment duration lasted for 9 days. Population density was estimated by cell count on a hemacytometer.

Results and discussion

Results and discussion The random balance design was used first to evaluate the main effect of 15 nutritive elements as mentioned previously plus the effect of H₃BO₃ at 2 levels of concentrations (+1) and (-1), which were chosen to express the highest and the lowest element concentrations used in the known artificial sea water media (ASP-2, ASP-6, ASP-12, ASM and Muller media). Data gained using this design showed that alga was tolerant to a wide range of macroelement concentration changes. Concerning micro- elements and H₃BO₃ the best algal yield was achieved at the levels given in table 2. For optimizing the major cations and anions concentrations in relation to algal yield the 2^{84} fractional factorial design was used, where the concentrations of the media on which the alga attained its maximum yield in the previous design was taken as a middle point in defining the (+1) and (-1) levels for this design. Results are given in table 1. After statistical treatment of the data we can conclude that algal yield was only significantly affected by the concentration changees of K. for this design. Results are given in table 1. After statistical treatment of the data we can conclude that algal yield was only significantly affected by the concentration changes of K, Mg, PO and CO₃. For optimizing their concentrations, a set of experiments was done using the steepest ascent method, where the composition of medium N° 15 on which the algal yield was maximum (table 1) was taken as original point. The highest mean algal yield (18 x106 cell/ml) was achieved on growth medium N°8. This is about 9-10 times greater than those recorded in the literature at about the same conditions of cultivation used in our experiments. In conclusion we are recommending a new medium for best *Dunaliella salina* growth as given in table 2.

Table I . 2⁸⁻⁴ fractional factorial design and steepest plan.

Factors	NaCl X ₁	кс1 Х ₂	MgCl ₂ X3	CaCl ₂	Na2SO	NaHCO3	NaNO3 X7	NaH_PO4 X8	is ity is				
Level mM/ -1 level +1 level 0 level Variation	1000 2000 1500		25 55 40	8 18 13	15 25 20	1.5 2.5 2.0	2 5 3.5	0.05 0.10 0.075	culture density after 9 days 10 ⁶ cell/ml				
unit (λ)	500	4	15	5	5	0.5	1.5	0.025	បន	-			- (3)
Experiment											042 0	CO10.	Salt
1 2 3 4	-1 +1 -1	-1 -1 +1	-1 -1 -1	-1 -1 -1	-1 +1 +1	-1 +1 +1	-1 +1 -1	-1 -1 +1	4.39 3.51 6.86			T ZIME	
4 5 6 7	+1 -1 +1 -1	+1 -1 -1 +1	-1 +1 +1 +1	-1 -1 -1 -1	-1 +1 -1 -1	-1 -1 +1 +1	+1 +1 -1 +1	+1 +1 +1 -1	4.56 6.97 4.61 4.63		3	1000	llaci
8 .9 10	+1 -1 +1	+1 +1 -1	+1 -1 -1	-1 -1 +1 +1	+1 -1 +1	-1 +1 -1	-1 +1 -1	-1 +1 +1	4.03 3.47 0.28 5.09	,		32	KC1
11 12 13	-1 +1 -1	+1 +1 -1	-1 -1 +1	+1 +1 +1	+1 -1 +1	-1 +1 +1	+1 -1 -1	-1 -1 -1	4.44 4.64 4.07	ŀ	1 1	Ē	Haci
14 15 16	+1 -1 +1	-1 +1 +1	+1 +1 +1	+1 +1 +1	-1 -1 +1	-1 -1 +1	+1 -1 +1	-1 +1 +1	4.59 7.66 5.17	d	ج م	18	2 CaCl ₂
Regre. coef.(b _i)	-0.23	0.5	0,46	-0.19	0.26	-0.46 -	-0.41	0.47					2 14
^ь iλ		2.0				-0.23		0.012			5	, t	N
	1000	16	55	18	15	1.5	2	0.100			. `		40
Level on path.										0.04	3	<u>م</u>	HaNo ₃
2 1 3 1 4 1 5 1	000 000 000 000	18 20 22 24 26	62 69 76 83 90	18 18 18 18 18	15 15 15 15	1.25 1.00 0.75 0.50 0.25	2 2 2 2 2 2 2	0.112 0.124 0.136 0.148 0.160	12.90 14.93 14.09 13.76 15.43 13.55	400	500 2 100	0.196	NaH2Po4
$\frac{7}{9}$ 1	000	28 30 32 34 36	97 104 111 118 125	18 18 18 18	15 15 15 15 15	0.00 0.00 0.00 0.00 0.00	2 2	0.172 0.184 0.169 0.208 0.220	13.55 14.49 18.10 15.91 16.01				
11 1 12 1 13 1	000 000 000 000	38 40 42	132 139 146	18 18 18 18	15 15 15	0.00 0.00 0.00	2 2 2	0.232 0.244 0.256	17.19 16.24 15.92				
	.000 .000	44 46	153 160	'18 18	15 15	0.00	2 2	0.268	16.17 17.24				

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111

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