TEMPORAL CHANGES IN CHEMICAL PROPERTIES OF A WARM CORE EDDY IN THE LEVANTINE BASIN OF THE EASTERN MEDITERRANEAN SEA

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Recently a persistent quasi-stationary warm-core eddy has been found south of Cyprus in the Levantine basin of the eastern Mediterranean Sea. A series of nine cruises over a three-year period (1989-1992) was carried out to examine its physical and chemical structure (BRENNER *et al.*, 1990; KROM *et al.*, 1992, 1993). The eddy is characterized by an isothermal, isohaline lens of water wedged between the seasonal and permanent thermoclines. In the winter, this thermostad extends from the surface to a depth of nearly 400 m, while in other seasons it lies in the layer from 200-400 m. Similarly, dissolved oxygen and nutrient (nitrate, o-phosphate and silicic acid) concentrations at the eddy core are essentially constant during winter, from the surface to a depth of 550 m. While during summer, the concentrations are constant at the depth interval 150-550 m.

Surface to a depit of 550 m, while during summer, the concentrations are constant at the depth interval 150-550 m. Temporal changes in temperature and salinity values at the eddy core indicated a process of renewal or replacement of the water trapped in the core of the eddy (BRENNER, 1993) and therefore we assumed that between February 1989 and March 1992 three different realizations of the Cyprus eddy were sampled. We examined the chemical characteristics of the core water during this time in order to check if the chemical parameters could be indicative of changes related to the proposed renewal processes. Each time when temperature and salinity increased, an increase in dissolved oxygen concentration followed, consistent with the fact that "new" water formed or flushed the eddy core (Table 1). During the period that the core was isolated from the surroundings, oxygen was utilized and the concentration decreased, increasing after the water renewal. It is therefore possible to see a cycle during the lifetime of a particular eddy. The increase in oxygen concentration indicates penetration of surface (or upper layer), oxygen rich water and not lateral penetration of water at the same depth as the core. Only during November 1989 we detected a deviation from this trend and saw a slight increase in dissolved oxygen. The changes in nitrate concentration were consistent with those noticed for the dissolved oxygen, but not as "unequivocal". The beginning of the eddy cycle was characterized by a decrease in nitrate concentration, followed by an increase during the time that the core was isolated from the surrounding water. An exception is the nitrate concentration found during September 1989, lower than the values found in May and November 1989.

nitrate concentration found during September 1989, lower than the values found in May and November 1989. Assuming that: a) the core water is isolated from the surroundings during the lifetime of the eddy; b) all the addition of nitrate to the core during the eddy's lifetime is due to decomposition of organic matter and c) surface water with negligible amounts of nitrate intrudes the core to form a new eddy, it is possible to compare the changes in dissolved oxygen and nitrate concentrations in the core of the eddy-09, was 8.1 µmole/kg (Table 1). Using Redfield's ratio of 138:16 for O₂:NO₃, the respective calculated amount of nitrate eddy-06 and Eddy-07. The difference in nitrate concentration actually measured was 0.86 µmole/kg, a very good fit. The same comparison was performed for Eddy-06 and Eddy-07. The difference in dissolved oxygen concentration measured was 4.8 µmole/kg, corresponding to a calculated value of 0.55 µmole/kg nitrate. The measured decrease in nitrate concentration was 0.47 µmole/kg, again in very good agreement with the calculated value.

Ortho-phophate concentrations were expected to follow nitrate concentration. As a whole it is true except for an unexplained increase during April 1990. However, one must keep in mind that the ortho-phosphate concentrations measured in the core of the eddy are very close to the detection limit of the method (KROM *et al.* 1992, 1993).

Date	Cruise	Temp.	Sal	O ₂	NO ₃	PO ₄	Si(OH)4
	no.	٥C	ppt		i µmol/kg	Sant	
5/88	02	16.44	39.08	222.8 (2.3)	0.55 (0.18)	0.008	1.55 (0.24)
2/89	03	16.43	39.15	220.0 (1.6)	0.57 (0.07)	<0.01	0.99 (0.06)
5/89	04	16.43	39.15	216.0 (0.9)	0.73 (0.10)	0.013 (0.004)	0.92
9/89	05	16.43	39.15		0.47	0.014 (0.009)	0.89 (0.27)
11/89	06	16.43	39.15	219.0 (1.6)	1.23	0.034 (0.015)	1.11 (0.36)
*4/90	07	16.68	39.27	223.8 (0.6)	0.76 (0.22)	0.051 (0.021)	1.36 (0.48)
10/90	08	16.68	39.27	214.3 (1.8)	1.02	0.021 (0.029)	1.29 (0.34)
*3/92	09	16.61	39.38	222.4 (1.2)	0.16 (0.05)	0.006	1.58 (0.06)

Table 1: Physical and Chemical characteristics of the eddy core (in parenthesis, standard deviation, * renewal of the eddy)

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